# [Time: 2.30 Hours]

### [Total Marks: 60]

Please check whether you have got the right question paper.

## NB:1. **All** questions are **compulsory**.

## 2. **Figures** to the **right** indicate **full** marks.

#### Q.1(A) Attempt any **two** of the following:

(a) Predict the major products of the following reactions.

LDA/THF

Br

ii) OOO  $2 NaNH_2$  ? Br

(b) Complete the following reaction and give its mechanism.

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(c) Complete the following reactions with suitable products.

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(d) Predict the products and give the mechanism of the following reaction.

COOEt

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(B) Attempt any **one** of the following:

(a) Predict the products and name the following reactions.

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(b) Explain Michael addition reaction and its mechanism.

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#### Q.2(A) Attempt any **two** of the following:

(a) Explain Von Richter rearrangement with its mechanism.

(b) What is Pummerer rearrangement? Explain its mechanism.

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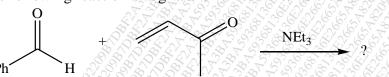
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- (c) Predict the product and name the following reactions.

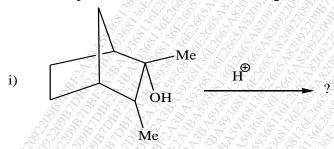
i) 
$$H_3C$$
 COOH  $\frac{HN_3}{H_2SO_4}$ ?

ii) Conc 
$$H_2SO_4$$

(d) Complete the following reaction and give the name. Write its mechanism.



- (B) Attempt any **one** of the following:
- (a) Explain Boulton Katritzky rearrangement with its mechanism.
- (b) Predict the product and name the following reactions. 4



ii) PhCH2CONHOCOCH3

#### Q.3(A) Attempt any **two** of the following:

Explain the following in IR Spectroscopy:

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- i)Vibrational Coupling ii)Study of hydrogen bonding
- (b) Explain 'donor-acceptor' interaction in nucleophilic addition reaction on
- 4 formaldehyde using FMO method. 4
  - Which of the following compounds will show a lower >C=O strectching frequency in IR spectroscopy and why?

i) 
$$CH_3$$
 And  $CH_3$   $CH_3$ 

- ii) COCH<sub>3</sub> COCH<sub>3</sub> And OH
- (d) 'Dimerization of ethane takes place via photochemical pathway'. Explain on the 4 basis of HOMO-LUMO interaction using  $\pi$  MOs.
- (B) Attempt any **one** of the following:
- (a) Draw the  $\pi$  MO diagram for the allyl anion and allyl cation. Show the interaction of their FMOs and predict the product of the reaction between the two ions.
- (b) Using IR Spectroscopy how can you distinguish between

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- i) Phenol and Cyclohexanol
- ii) Ethylbenzene and o-xylene
- Q.4(A) Attempt any **two** of the following:
  - (a) Write the fragmentation pattern of the following molecules in Mass spectroscopy;
    - i) Ethyl benzene ii) 3-pentanone
  - (b) Explain the following terms in NMR spectroscopy with suitable example.i) Long range coupling ii) Coupling constant
  - (c) An Organic Compound having M.F.  $C_4H_8O_2$  showed following spectral dada: I.R. = 2970 cm<sup>-1</sup>, 1745 cm<sup>-1</sup>, 1200 cm<sup>-1</sup>.  $^{13}\text{CNMR} = \delta$  14 (q),  $\delta$  32 (q),  $\delta$  82 (t) and  $\delta$  185 (s).
  - Deduce the structure of the compound and justify your answer.

    (d) Write the number of signals and splitting pattern observed in the NMR spectra of 4 following molecules:

- (B) Attempt any **one** of the following:
- (a) An Organic Compound having M.F.  $C_{10}H_{12}O$  showed following spectral dada: **I.R.** = 3090 cm<sup>-1</sup>, 2970 cm<sup>-1</sup>, 1700 cm<sup>-1</sup>, 850 cm<sup>-1</sup>.

<sup>1</sup>**H-NMR** = δ 0.9 (3H, t), δ 4.3 (2H, q), δ 7.3 (2H, d, 7.5Hz), δ 7.6 (2H, d, J = 7.5Hz) and δ 2.3 (3H, s ).

<sup>13</sup>**CNMR** =  $\delta$  15,  $\delta$  30,  $\delta$  35,  $\delta$  110,  $\delta$ 120,  $\delta$  135,  $\delta$ 140 and  $\delta$  210.

Deduce the structure of the compound and justify your answer.

- (b) Explain the following in Mass spectroscopy with suitable example;
  - i) McLafferty rearrangement ii) Ortho effect
- Q.5 Attempt any **four** of the following:
  - A What is Tandem alkylation? Explain with suitable example.

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B Predict the products and give the mechanism of the following reaction.

- C Explain Passerini reaction with its mechanism.
- D Complete the following reactions.

- E Discuss hard and soft electrophiles? Give one example of each. Comment on the relative energies of their LUMOs.
- F Calculate the  $\lambda$ max of the following compounds 3

- G Explain the term D<sub>2</sub>O exchange and its application in NMR spectroscopy.
- H Explain the term isotopic abundance and Molecular ion peak in Mass

  Spectroscopy.

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