Msc. II - Sem. IV - oct. 2016 Organic Chemistry - paper IV

QP Code: 76845

	o .	(21/2	Hours)		[Total Marks : 6
N.B. :	(1) (2)	All questions are compulsory Figures to the right indicate	[/] . maximum	marks	
(a)	Answer any two of the following:				
	(i)	Why are drugs converted	into pro	drugs? Give the typ	pes and ideal
	(ii)	Explain 'QSAR' and discu	iss how th	ne Taft equation he	lps to prodict
	(iii)	the cheet of stelle factors on the biological activity of a diffs.			
	(iv)	Explain the studies carr relationship of structure t	ied out b	y Hansch for Qua	
	*	forms of modified Hansch	equation	ıs.	
(b)	(b) Answer any one of the following:				
		 (i) Give the synthesis and (I) Oxyphenbutaz (ii) Give the synthesis and (I) Methotrexate 	zone	(II) Fenofibrate	;
2. (a)	() () () () () () () () (
	 (i) Starting from enolic interprediate explain the role of lipother the conversion of coenzyme A to acetyl coenzyme A. (ii) Write a note on biomodelling giving any one example. 				
	(iii)	Give the plausible mechanism for the conversion of methyl malonyl coenzyme A to succinyl coenzyme A brought about by coenzyme			
٠	B ₁₂ dependent enzyme. (iv) Match the following coenzymes with their metabolic functions/ structural reatures.				
	I)	NAD	A)	Isoalloxazine ring	
	II)	Ryridoxal phosphate	B)	Carrier of CO ₂	
3	III)	Biotin	C)	Niacin	
	IW	FAD	D)	Racemisation	•

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(b) Answer any one of the following:

- Give the mechanism of conversion of pyruvate to acetolactate by acetolactate synthase which requires thiamine pyrophosphate as a coenzyme.
- JPD 16088 DE 5088 Give the name and the structure of the coenzyme involved in the (ii) following enzymatic biochemical transformations.
 - Tryptophan to serotonin I)
 - II) Acetyl coenzyme A to malonyl coenzyme A
- 3. (a) Answer any two of the following:
 - Write a note on production of vitamins by fermentation (i)
 - (ii) Give any two examples of each of the following encyme catalyzed reactions.
 - I) Hydroxylation
 - II) Reduction
 - Giving examples illustrate how microbial Pansformation is used in (iii) organic synthesis.
 - How are enzymes immobilised by cross linking? What are the (iv) advantages of immobilization of engines.
 - (b) Answer any one of the following:
 - Write a note on production 6-aminopenicillanic acid using an (i) enzyme in immobilzed form.
 - Explain the structural features of glycogen. (ii) Explain the role of phosphoglucomutase in glycogen breakdown.
- 4. Answer any one of the following:
 - What is the importance of 'ultra sound' in green synthesis? What are (i) the required conditions, for the use of ultra-sound in green synthesis? Explain how Cannizzaro reaction is carried out under sonication.
 - With reference to green chemistry write notes on : (ii)
 - Ionic Liquids
 - Solid supported green synthesis.

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- Name and give the structure of any two green solvents. Give their (iii) green advantages. How is Diels-Alder reaction carried out, using water as a solvent?
- (iv)
- (b) Answer any one of the following:
 - (i)
- any one of the following:

 For the synthesis of para aminodiphenylamine, compare the conventional method of synthesis with the green methodo.

 Give the traditional process and the green process of adipic acid and give the advantages. (ii)
- 5. Answer any one of the following:
 - Write a note on the concept of 'Soft drugs' What are the properties of (a) soft drugs?
 - Give the synthesis and one application Plabetalol. (b)
 - Explain the stereospecificity observed in the oxidation of an alcohol (c) by alcohol dehydrogenase which uses NAD+ as a coenzyme.
 - Show how an amino acid is converted to a keto acid by pyridoxal (d) phosphate.
 - Giving examples show how chiral hydroxy acids are prepared by (e) enzymatic processes.
 - Name the enzymes involved in glycogen synthesis. (f)
 - What is it important to avoid 'waste', from a green perspective? Give (g) two reactions that have 100% atom economy.
 - (h)