MSC-I - Sem-II - Oct 2016-Chemistry- Paper-II **QP Code: 75798** (Inorg) (2½ Hours) [Total Marks: 60 All questions are compulsory. Figures to the right indicate full marks. "017/27/27/27/25. Attempt any two of the following:-By applying the concept of hybridisation, derive the wave function for hybrid orbitals of methane molecule. What are electron deficient species? On the basis of LCAO-MO concept, explain the bonding in diborane. Discuss the different types of hydrogen bonding with suitable examples. Define magnetic susceptibility. How is it determined by Faraday's method? (b) Attempt any one of the following:-Derive the wave function for hybrid orbitals of BeH, molecule considering only sigma bonding. On the basis of LCAO-MO method, discuss the bonding in an-iodide ion. 4 2. (a) Answer any two of the following:-Explain Abelian and non abelian point group with switable examples. 4 Discuss the criteria for a set of elements to form Sgroup by giving suitable 4 examples. What are irreducible representations? State their characteristics. 4 Discuss the application of Group Theory, Explain the sigma and pi bonding 4 in AB, molecule. (b) Attempt any one of the following:-Construct the group multiplication table for C2v point group. 4 Derive the matrix representation for identity operation. 4 3. (a) Attempt any two of the following Discuss the sources, biochemical effects and toxicity of Mercury. 4 Explain the harmful effects of radioactive pollution on human cells. 4 Give the classification and origin of geothermal energy. 4 Elaborate on wind power and hydro power as indirect sources of energy. 4 (b) Attempt any one of the following:-Discuss the toxic effects and treatment of Cadmium poisoning. 4 Write a note on generation of power from nuclear energy. 4

N.B. : (1)

1. (a)

(2)

(i)

(ii)

(iii)

(iv)

(i)

(ii)

(i)

(ii)

(iii)

(iv)

(i)

(ii)

(i)

(iii)

(iv)

(i)

(ii)

' (ii)

TURN OVER

11- 8-9-19 - 2(K/07/10)

(a) Attempt any two of the following:-

- Comment "Co-operative binding of oxygen is observed in haemoglobin (i) molecule".
- Explain the enzymatic reactions catalysed by cytochrome P-450. (ii)
- Discuss the composition and the function of transferrin and ferritin. (iii)
- (iv)
- (b) Attempt any one of the following:-
- Discuss the conversion of atmospheric nitrogen into ammonia by nitrogenase 4

 Explain the various steps in the reaction catalysed by superoxide dismutation (i)
 - Explain the various steps in the reaction catalysed by superoxide dismutase.

 y four of the following:
 Explain ion-dipole and London forces with suitable examples. (ii)
- 5. Attempt any four of the following:-
 - (a)
 - State and explain Curie-Weiss Law for antiferromagenetic substances. (b.)
 - Define subgroup. Give the characteristics of subgroup. (c)
 - Explain symmetry restrictions on the dipole moment in smolecule. (d)
 - Describe the toxic effects and treatment of poisoning by Copper. (e)
 - Write a short note on production of electricity from solar energy. (f)
 - Discuss the structural features and function of hermocyanin. (g)
 - Explain the reactions of peroxidase using a suitable example. (h)

12