## (2 1/2 Hours)

[Total Marks: 60

N.B.: (1) All questions are compulsory.

- (2) Use of log table or nonprogrammable calculator is permitted.
- Q.1 (a) Attempt any Two of the following:

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- (i) What are the precautions to be taken while working in the chemical laboratory?
- (ii) What do you understand by the term hazardous codes? Give symbols of any two hazardous codes.
- (iii) Explain various steps involved in the process of patenting.
- (iv) What are chemical standards and reference materials? What is their importance in chemical analysis?
- (b) What is the importance of the first aid? What first aid treatment is to be given for the burns due to hot concentrated acid?

OR

- (b) What are the basic requirements to be fulfilled before applying for the patent?
- Q.2 (a) Attempt any Two of the following:

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- (i) Discuss the principle and applications of microfiltration.
- (ii) Describe the technique of ultrafiltration with suitable example. How does it differ from other techniques?
- (iii) What is reverse osmosis? With the help of neat labelled diagram, explain how is it used for the purification of brackish water.
- (iv) Differentiate between dialysis and electrodialysis with respect to separation mechanism.
- (b) Discuss the advantages of automatic analysis.

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- (b) Describe the operation of Flow Injection Analysis system for the determination of chloride in water.
- Q.3 (a) Attempt any Two of the following:

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- (i) Explain the concept of Green Chemistry. Why is it mandatory to adopt green chemistry?
- (ii) Explain electrochemical synthesis with suitable example.
- (iii) Why super critical fluids are preferred as solvents in green chemistry?
- (iv) What are photochemical reactions? Give the advantages of photochemical reactions.

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			Why is atom economy calculated?  N-ethyl propanamide is synthesized using the following reaction. Calculate atom economy for it in percentage.  CH <sub>3</sub> CH <sub>2</sub> COOC <sub>2</sub> H <sub>5</sub> +C <sub>2</sub> H <sub>5</sub> NH <sub>2</sub> → CH <sub>3</sub> CH <sub>2</sub> CONHC <sub>2</sub> H <sub>5</sub> + C <sub>2</sub> H <sub>5</sub> OH  (Atomic weights: H=1, O = 16, C = 12, N=14)	a)
			OR .	
		(p)	What do you understand by the terms Inherently Safer Design (ISD) and Process Intensification (PI)? Describe in details with example.	4
				8
Q.4	(a)	(i	mpt any Two of the following:  i) What is meant by electrophoretic and electro osmotic flow in zone electrophoresis? Describe the basic instrumentation of this technique.  i) What are the detectors used in capillary electrophoresis? Explain any one in	,
			detail.  i) Discuss the principle and applications of SDS page and iso-electric	
		(iì	a) Discuss the principle and applications of the first of the principle and applications of the first of the principle and applications of the	•
		(is	focusing.  v) Explain miscellar electrokinetic capillary chromatography.	
	(p)	Hou	v are nano materials classified? What are the different analytical niques used for their characterization?  OR	4
	<b>\</b>	~~ r t	a suitable block diagram explain the working of gel electrophoresis.	4
	<b>(b)</b>	Witi	a suitable block diagram oxpress	
			any Four of the following:	12
Q.5	Ati		The same about he taken while storing the inginy initialination indicates	
		(i)		
		(ii) (iii)	What is pharmacopoeia? How is it important in order of glucose in blood.  Discuss the applications of multilayer films in analysis of glucose in blood.	
		(iv)	What are gas monitoring equipments?	
		(.v)	Define the terms:	
		()	(a) Sustainable development (b) Zero Waste	
		(vi)		
		(vii)	Give an account of various supporting media used in zone electrophoresis.	
		viii)	Give an account of various supporting media association of the particle Give the instrumental methods used for the determination of the particle	
			size of nano material.	