

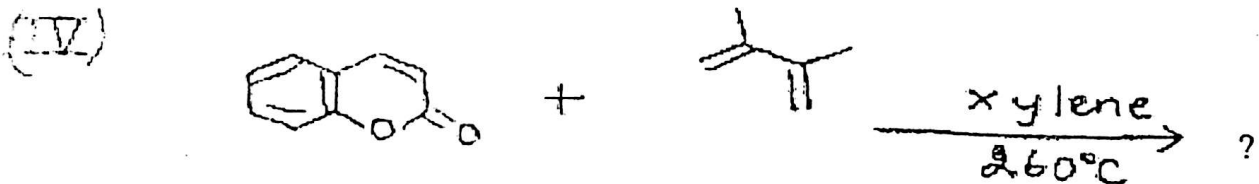
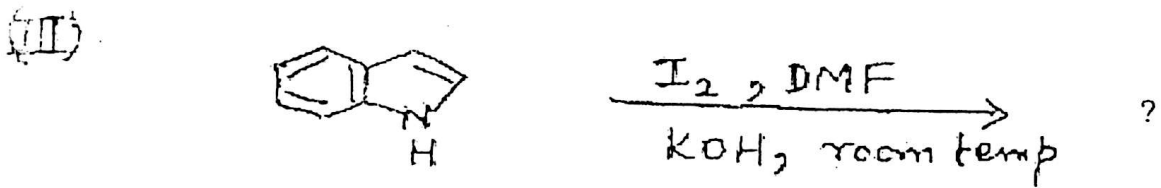
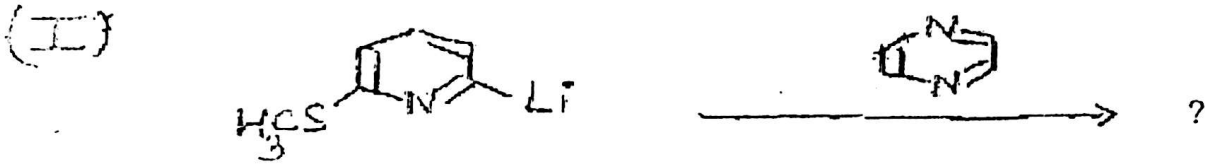
(2½ Hours)

- NOTE: (1) All questions are compulsory.
 (2) Figures to the right indicate full marks.

Q.1) Answer any two of the following :—

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- (i) Complete the following reactions :—



- (ii) (I) Give the synthesis of pyrimidines from
 (A) 1,3 - dicarbonyl compound
 (B) 1,3,5 - triazine
 (II) Give reason : The diazines show more resistance to electrophilic substitution than pyridine.
 (iii) Give two methods of synthesis of isoquinoline.

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(iv) (I) Give Traube synthesis of purines.

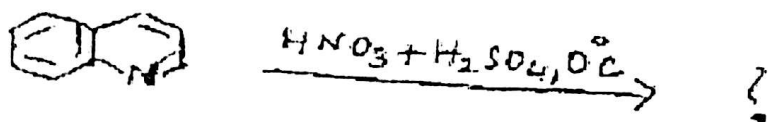
(II) What is the action of the following reagents –

(A) CH_3I , CH_3OH , 100°C on purine,

(B) $\text{H}_2\text{N}-\text{NH}_2$, room temp. on 6-chloropurine?

(v) Answer any one of the following:—

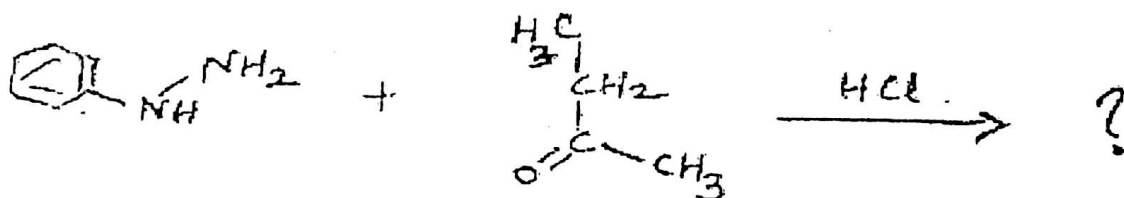
(i) Complete the reaction given below:—



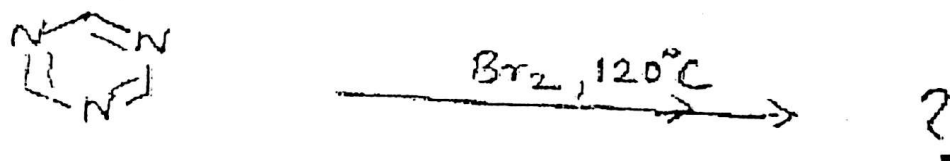
Give the mechanism and explain the regioselectivity.

(ii) Complete the following reactions.

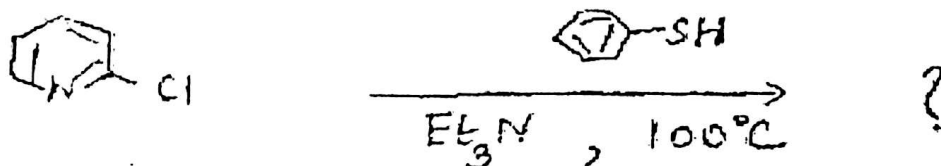
(I)



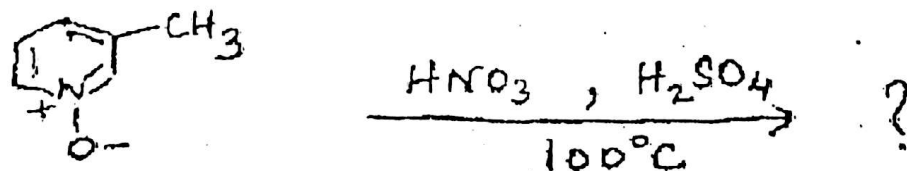
(II)



(III)



(IV)



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(a) Answer any two of the following :—

- (i) Discuss the occurrence, biological role and structural features of corticosteroids.
- (ii) Write a note on steroidal alkaloids.
- (iii) Draw the structure of testosterone. How is 16-DPA synthesised from a plant sapogenin?
- (iv) Give the synthesis of androsterone from 16-DPA.

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(b) Answer any one of the following :—

- (i) How is 16-DPA converted to oestrone?
- (ii) Give the structure of 5α -cholanic acid. How will you synthesise jasmolone?

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(c) Answer any two of the following :—

- (i) (I) How are vitamins classified?
(II) Give the synthesis of Vitamin B₂ from 3, 4-dimethylaniline and D (-) ribose as starting materials.
- (ii) (I) State the sources and biological importance of Vitamin B₁₂.
(II) Outline the synthesis of Vitamin B₁.
- (iii) State the biological importance and give the synthesis of—
(I) α -tocopherol
(II) Vitamin K.
- (iv) Give the synthesis of penicillin-G from D-penicillamine.

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(d) Answer any one of the following :—

- (i) Discuss the structure of Cephalosporin C based on its degradation studies.
- (ii) Briefly describe the sources, biological importance and the synthesis of Vitamin B₆.

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(e) Answer any two of the following :—

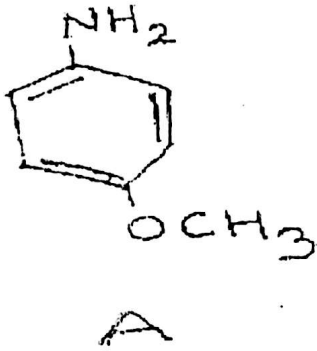
- (i) Discuss off-resonance decoupling in ¹³C-NMR spectra of any two compounds.
- (ii) Sketch the proton decoupled spectrum, DEPT - 135, DEPT - 90 and DEPT - 45 of the compound (2-methyl-1-propanol) using ¹³C-NMR value δ ppm: 69.3, 30.7 and 18.7.
- (iii) Explain HETCOR technique with a suitable example.
- (iv) Sketch the COSY spectrum of ethyl acetate.

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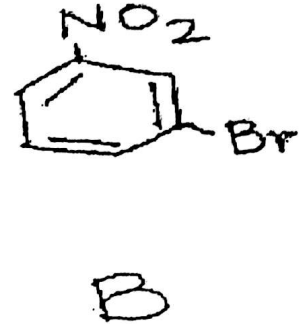
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(c) Answer any one of the following :—

- (i) What is ESR spectroscopy? Discuss its principle. State the applications of NMR spectroscopy in the field of medical science.
- (ii) Assign ^{13}C -NMR Chemical shift to all the aromatic carbons using the chemical shift correlation table given below, for the compounds A and B.



and

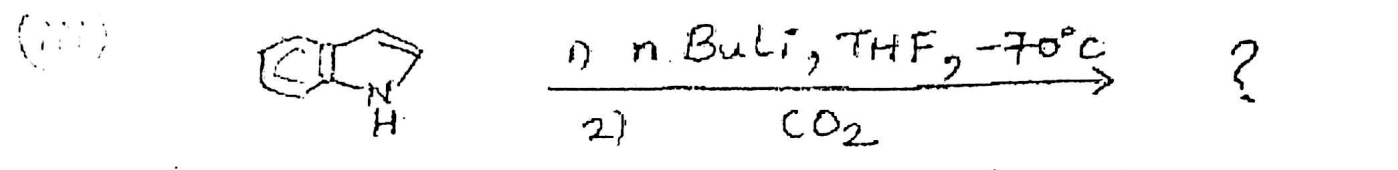
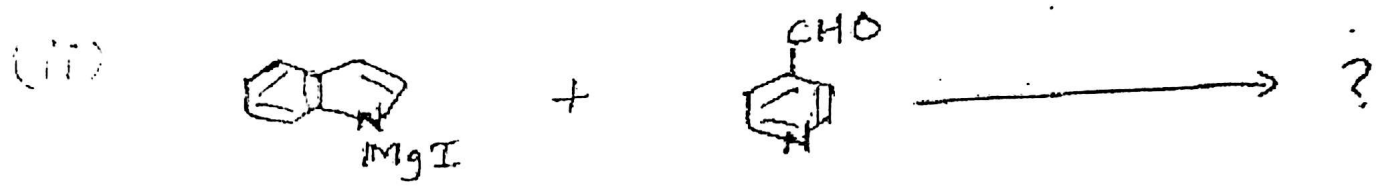
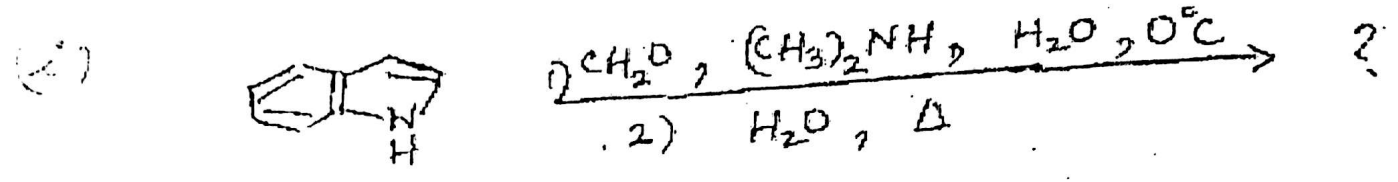


Substituent	Increments in ppm			
	ipso	ortho-	meta-	para
-NH ₂	19.2	-12.4	1.3	-9.5
-OCH ₃	31.4	-14.4	1.0	-7.7
-NO ₂	19.6	-5.3	0.9	6.0
-Br	-5.4	3.4	2.2	-1.0

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5. Answer any four of the following :—

(a) Complete the following reactions :—



- (b) Give any two methods of synthesis of coumarin.
- (c) Give the synthesis of muscone.
- (d) How is 16-DPA converted to progesterone ?
- (e) State the sources and biological properties of rotenoids. Draw the structure of rotenone.
- (f) Give the synthesis of Pyrethrin-I
- (g) The following chemical shifts were observed in the ¹³C-NMR of p-methyl anisole, δ 20, 55 and 157 ppm. Match the chemical shifts with the appropriate carbons. Justify your answer.
- (h) Discuss the applications of fluorescence spectroscopy.

VF-Con. 6085-15.