(2½ Hours)

[Total Marks:60

N.B.: (1) All questions are compulsory

- (2) Figures to the right indicate full marks.
- (A) Answer any two of the following:-
 - (a) Draw the MO diagrams for the allyl cation and allyl anion. Show the interaction of their FMOs and predict the product of the reaction between the two ions.
 - (b) Compare the aromaticity of [10] annulene and [18] annulene.
 - (c) Draw the Frost Musulin diagram for cyclooctatetraene showing the distribution in the MOs. Comment on the shape of the molecule and its aromaticity.
 - (d) Answer the following:-
 - (i) What are antiaromatic compounds? Give an example.
 - (ii) Explain the use of ¹H-NMR to prove aromaticity of a compound.
- (B) Answer any one of the following:-
 - (a) Explain the significance of the HOMO-LUMO gap in the UV 4 absorption of ethene and but addiene with π MO diagrams.
 - (b) Draw the Frost Musulin diagrams for the tropylium cation and cyclobutadiene. Show the distribution of electrons in their MOs and comment on their aromaticity.
- 2. (A) Answer any two of the following:
 - (a) Complete the following reaction and explain its mechanism with stereochemistry.

- (b) Explain the use of the following techniques in determining mechanism 4 of the reactions specified.
 - (i) Cross-over experiments in rearrangement reactions.
 - (ii) Detection and trapping of intermediates in aromatic S_N on chlorobenzene.
- (c) Arrange the following in increasing order of acidity and justify your answer.

Benzoic acid, benzyl alcohol, phenol, p-chlorobenzoic acid.

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(d) Answer the following :--

- (i) Compare the basicity of p-toluidine and p-chloroaniline
- (ii) 'Piperidine is more basic than pyridine'. Explain.
- (B) Answer any one of the following:-
 - (a) With the help of an energy profile diagram explain:
 - (i) Hammond's postulate
 - (ii) Transition state theory.
 - (b) Answer the following:-
 - (i) Predict the products of the following reaction and comment on their stabilities.

$$CH_2 = CH - CH = CH_2 + HB_{\gamma} - \frac{80^{\circ}C}{40^{\circ}C} ?$$

- (ii) Write a complete equation to represent Chugaev reaction.
- 3. (A) Answer any two of the following:
 - (a) Explain 'atropisomerism' with reference to biphenyls. Write the structures of a pair of enantiomeric biphenyls with their configurational descriptors.
 - (b) Answer the following:-
 - (i) Explain 'improper' axis of symmetry in a molecule with a suitable example.
 - (ii) Differentiate between an asymmetric and a pseudoasymmetric centre in a molecule with suitable examples.
 - (c) Explain the substitution and symmetry criteria for differentiating between homotopic and enantiotopic ligands with suitable examples.
 - (d) Explain enantiomerism in the following compounds with suitable 4 examples.
 - (i) N oxides
 - (ii) Phosphines

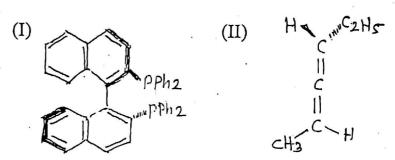
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(B) Answer any one of the following:

(a) Answer the following

(i) Assign configurational descriptors to the following:



- (ii) Write the structure of a molecule with enantiotopic faces and assign stereochemical descriptors to these faces.
- (b) Explain the principle of planar chirality with reference to cyclophanes. Write the structure of a chiral paracyclophane with its stereochemical descriptor.

4. (A) Attempt any two of the following:-

(a) Predict the products of the following reactions:

(i)
$$CH_3 = NOH NOBH_3CH$$
 ?

(iii)
$$CH_3 \xrightarrow{\text{1) } O_3} ?$$

(iv)
$$C_6H_5-\ddot{U}-cH_2-cH_2-c$$
 Se02 ?

- (b) Give the general mechanism of metal hydride reduction. Illustrate the chemoselectivity of NaBH₄ with two examples.
- (c) Complete the following reactions and name them.

(i)
$$CH = CH - CH - CH_3 \xrightarrow{Al(och Me_2)_3} q$$
(ii)
$$CF_3 CO_3 H$$

$$Q$$

(ii)
$$\frac{CF_3CO_3H}{CF_3COOH}$$
 ?

(d) What is Swern oxidation? Explain its mechanism with a suitable example.

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- (B) Answer any one of the following:-
 - (a) Complete the following reactions and give the mechanism of any one:

(b) Suggest suitable reagents for the following:

$$(i) \qquad \stackrel{\text{OH}}{\longrightarrow} \qquad \stackrel{\text{P}}{\longrightarrow} \qquad \stackrel{\text{P}}{\longrightarrow$$

$$(iii) \qquad \stackrel{\text{CH3}}{\longrightarrow} \qquad \stackrel{\text{CH0}}{\longrightarrow}$$

(iv)
$$CH_3-CH-OH$$

$$CH_3-CH-OH$$

$$CH_3-CH-OH$$

$$CH_3-CH-OH$$

- 5. Answer any four of the following:-
 - (A) Explain whether the followings ions are aromatic or not: cyclopropenyl cation, cyclopropenyl anion, cyclopentadienyl anion.
 - (B) Comment on the aromaticity of pyridine and furan.

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(C) The rate of elimination of HBr from compound A is 1.4 times faster than from compound B. What is the type of kinetic isotope effect displayed and how is the mechanism (E₁ or E₂) proved?

- (D) What is specific acid catalysis? Explain its mechanism with a suitable example.
- (E) Write the structure of threo-2,3 dibromo butanoic acid and assign configurational descriptors (R/S) to the chiral centres in the molecule.
- (F) Assign configurational descriptors to the top and bottom faces of the following:

(G) What is Raney Nickel? Predict the products X and Y in the following reaction:

$$CH_{3}CH_{2}CH_{2}CHO \xrightarrow{HSCH_{2}CH_{2}SH} X \xrightarrow{Raney Ni} Y$$

- (H) Illustrate the use of the following in organic synthesis with one example each.
 - (i) PCC
 - (ii) Chloranil
 - (iii) Dess-Martin reagent

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